Application of Radium Isotopes to Study the Transport of Dissolved Inorganic Carbon from the Caloosahatchee River to the Gulf of Mexico



INTRODUCTION

- Dissolved Inorganic Carbon (DIC) in the marine carbonate system (CO₂, H₂CO₃, HCO₃⁻ and CO₃²⁻) acts as a major buffer in the ocean making it integral in the study of ocean acidification.
- •Alkalinity refers to water's ability or inability to neutralize acids. It determines the buffering capacity of seawater and moderates the change in pH.
- •DIC is an important component of alkalinity (HCO₂⁻ and CO_{2}^{2}).
- •Short-lived radium isotopes (²²³Ra & ²²⁴Ra) can be used to track the residence times, mixing rates, transport fluxes of water and associated DIC as well as nutrients in the coastal marine environments.

OBJECTIVE

• Use radium isotopes to track the transport of DIC and nutrients from the Caloosahatchee River to the Gulf of Mexico.

RADIUM

METHODOLOGY

- 40 liters of water was collected in two 20 L cubitainers passing through a 1.0-µm cartridge filter & then gravity filtered through Mn fibers.
- The manganese oxide coated fibers were analyzed using a Radium Delayed Coincidence Counter (Radecc) for ²²³Ra and ²²⁴Ra isotopes.

DISSOLVED INORGANIC CARBON

- 125 mL of water was filtered through a 0.45-μm syringe filter and spiked with a 5- μ l HgCl₂ solution.
- The samples were run in the total organic carbon analyzer, TOC-L, and then were calculated to find the DIC concentration.

ALKALINITY

- 125 mL of water was filtered through a 1.0-μm cartridge filter.
- The Easy Plus automated titrator was used to perform a titration and then the data was used to calculate alkalinity, expressed as mg CaCO₂/L.

NUTRIENTS

 125 mL of water was collected and analyzed using an auto-analyzer.

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BROADER IMPLICATIONS

• Due to the environmental problems in Southwest Florida and a lack of research regarding DIC and nutrient transport, this study will help determine the importance of freshwater plume discharge in transporting DIC, and its affect on the Gulf.



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